No	Information of Engineering Materials I (2018-2019)	
1	Unit name:	Concept of Materials Science
2	Code:	Met- 31023
3	Classification:	Engineering subject
4	Credit value:	2.5
5	Semester/ Year Offered:	1/3
6	Pre-requisite:	
7	Mode of delivery:	Lecture, Tutorial, Assignment
8	Assessment system and breakdown of	
	marks:	
	Test	30%
	Mid-term/ final Examination	70%
9	Academic staff teaching unit:	1
10	Course outcome of unit:	
	In this course, students will be able to	
	- explain main type of engineering materials (Metals, Ceramics, Polymer, Semiconductor	
	and Composite)	
	- explain Atomic Bonding (Metallic bonds, Covalent bond, Ionic bond and Vander Waals	
	bonds)	
	- explain and apply Lattice, Basis, Units Cells and Crystal Structures to calculate the	
	problems.	
11	Synopsis of unit:	
	The course describes Simple cubic, Body-centered cubic, Face-centered cubic and Hexagonal	
	Close-Packed structure diagrams. Type of diffusion problems(Fick's first Law and Fick' second	
	Law.	

12	Topic:	
	Chapter 1	
	Classification of Materials	
	Classification of Materials Based on Structure	
	Environmental and Other Effects	
	Materials Design and Selection	
	Chapter 2	
	The structure of the atoms	
	Atomic Bonding	
	Binding Energy and Interatomic spacing	
	Chapter 3	
	Lattice, Basis, unit Cells and Crystal Structures	
	Points, Direction and Plane in the Unit Cell	
	Crystal Structures of IonicMaterials	
	Chapter 4	
	Point Defects	
	Other point Defect	
	Dislocations	
	Schmid'law	
	Chapter 5	
	Application of Diffusion	
	Rate of Diffusion	
	Mechanisms for Diffusion	
	Applications of Diffusion	
	Applications to Polymers	
14	Main Reference; The Science and Engineering of Materials, Six Edition, Donald R. Askeland	
	The second and Engineering of Fraction, Dir Edition, Donard R. Abkeland	
	-	
15	Additional references: - : Materials Science and Engineering An Introduction, Eight Edition,	
	William D. Callister, Jr. David G. Rethwisch	